

Table S1 Textural parameters of SNCH and DNCH.

Samples	Specific surface area (m ² /g)	Pore diameter (nm)	Pore volume (cm ³ /g)
SNCH	41.5	38.6	0.11
DNCH	74.3	28.4	0.26

Table S2 Parameters of Langmuir and Freundlich isotherm equations.

Adsorbents	Temperature	Langmuir			Freundlich		
		q _m (mg/g)	K (mg/L)	R ²	n	K _F (mg/g)	R ²
SNCH	298K	146.20	27.43	0.9991	5.46	47.30	0.9738
	308K	105.26	30.18	0.9992	5.53	33.58	0.9766
	318K	77.10	36.54	0.9991	5.37	23.35	0.9801
DNCH	298K	226.76	20.18	0.9990	5.8665	82.04	0.9686
	308K	206.19	24.50	0.9991	5.4016	67.23	0.9663
	318K	186.22	28.46	0.9991	5.4484	59.54	0.9658

Table S3 Thermodynamic parameters of phenol adsorption by SNCH and DNCH.

Adsorbents	Temperature	ln K	ΔG^0 (kJ/mol)	ΔH^0 (kJ/mol)	ΔS^0 (J/mol K)
SNCH	298 K	4.37	-10.827		
	308 K	4.02	-10.294	-36.279	-85.206
	318 K	3.41	-9.016		
DNCH	298 K	5.43	-13.453		
	308 K	4.74	-12.138	-34.012	-69.800
	318 K	4.53	-11.977		

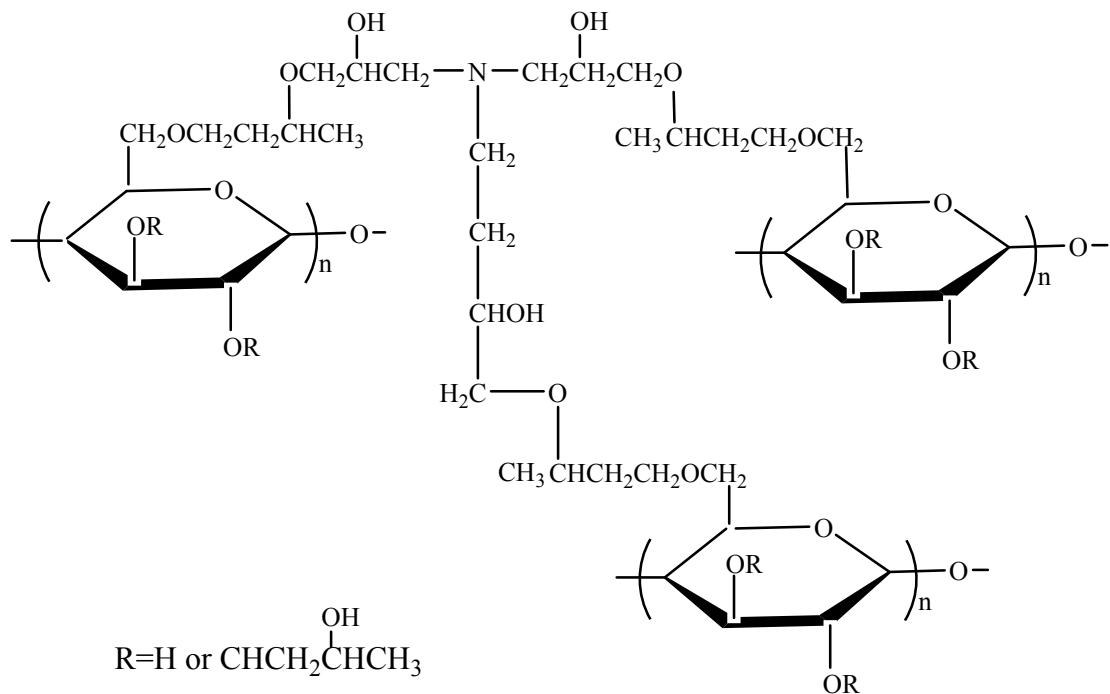


Fig. S1. Structure of the crosslinked HPC.

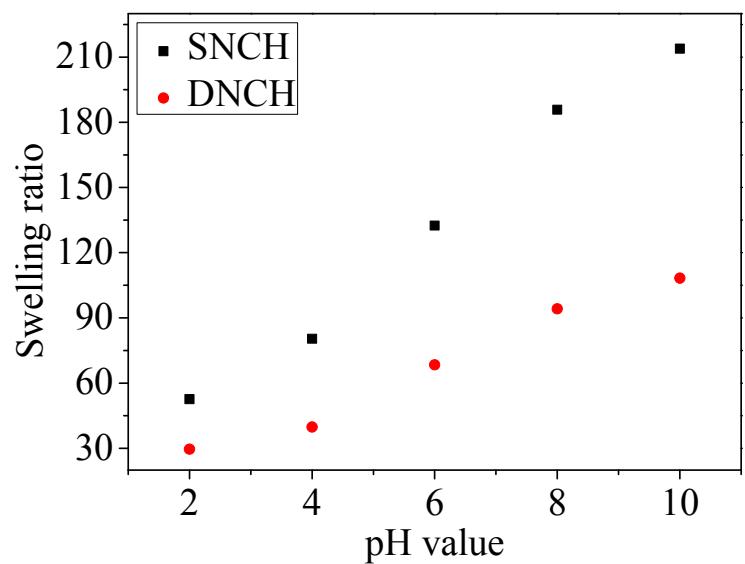


Fig. S2. Swelling ratios of the SNCH and DNCH at different pH values.

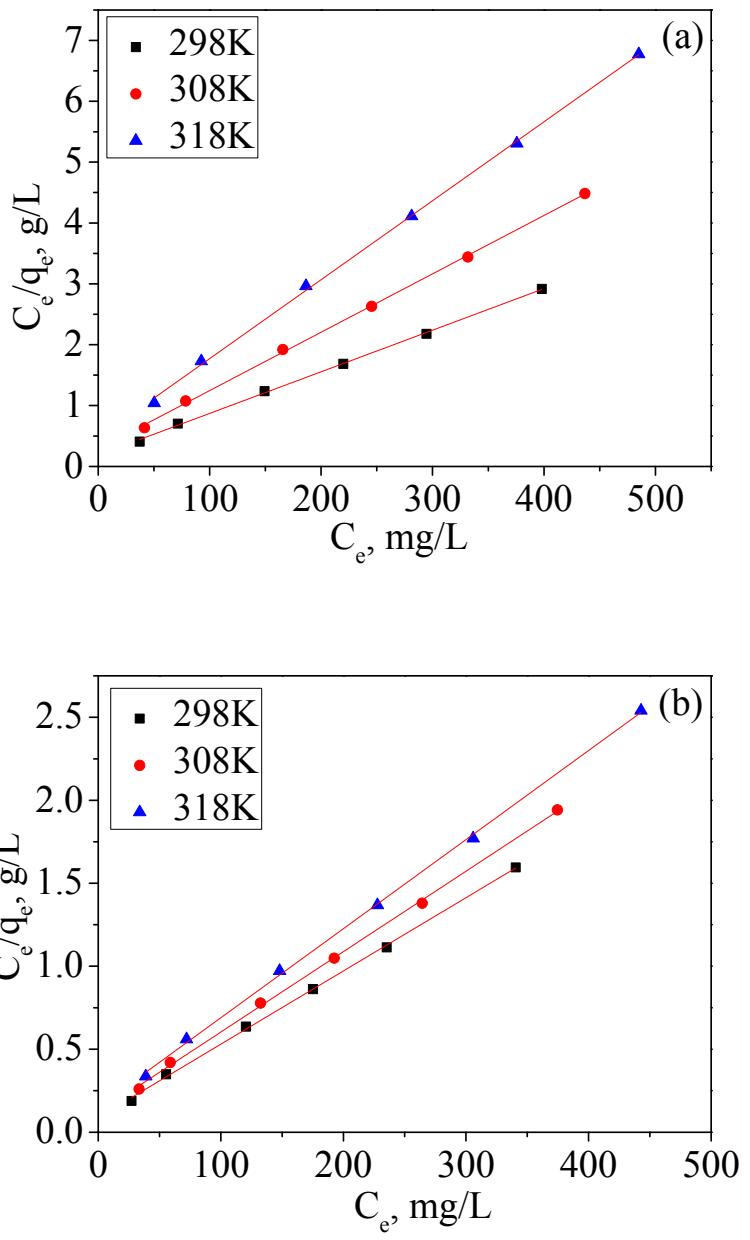


Fig. S3. Adsorption isotherm data presented in terms of the linearized Langmuir model for (a) SNCH and (b) DNCH.

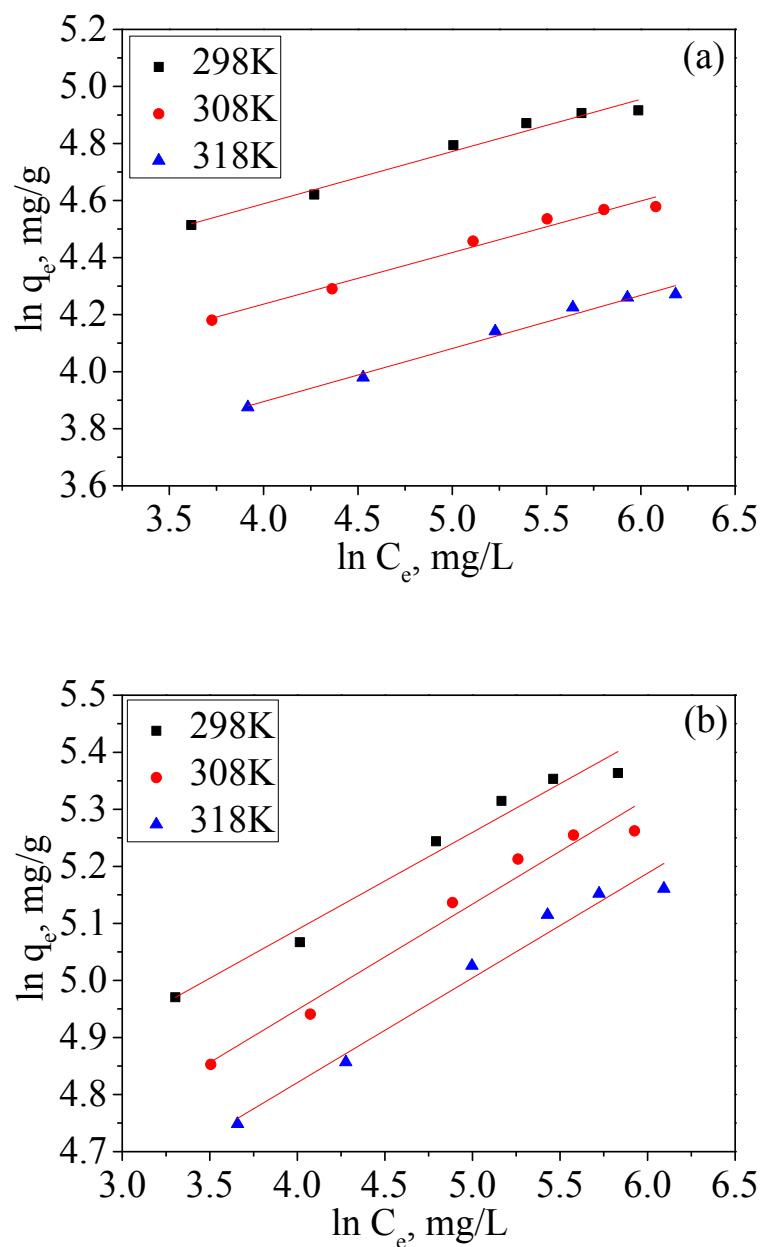


Fig. S4. Adsorption isotherm data presented in terms of the linearized Freundlich model for (a) SNCH and (b) DNCH.

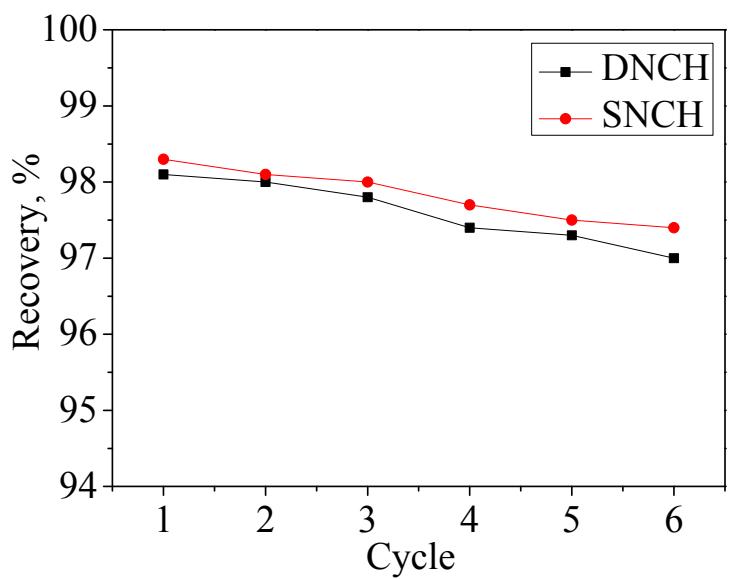


Fig. S5. Effect of reused times on the adsorption capacity of phenol.